# **Notiz** / **Note**

# **Gas-Phase Generation and Characterization of Hydrogen Cyanide N-Methylide**

### **Norman Goldberg, Muhammad Iraqi, and Helmut Schwarz"**

Institut für Organische Chemie der Technischen Universität Berlin, StraDe des 17. Juni 135, D-10623 Berlin, **F.** R. G.

Received July 23, 1993

**Key Words:** Ylides / Distonic ions / Neutralization-reionization mass spectrometry

The recently described (Maier et al.<sup>[2]</sup>) title compound is not only viable in an argon matrix (10 K) but is also accessible in the gas phase. Electron impact ionization of  $(CH_3)_3CCX_2NC$ 

The increasing importance and conspicuous properties **of** distonic ions<sup> $[3]$ </sup> - species in which the charge and radical centres are formally separated - **as** central intermediates in areas as diverse as chemistry in the gas phase or in frozen matrices are well documented in several review articles<sup>[4]</sup>. An interesting sub-class are the a-distonic cations which formally correspond to the ionized forms of ylides f51. **A** typical example which highlights some of the unusual features of these molecules is  $H_2\dot{C}-\dot{O}H_2$  (1). This radical cation has indeed remarkable properties in comparison with its conventional isomer, the methanol molecular ion CH<sub>3</sub>OH<sup>+</sup> (2) (Scheme 1). In contrast to the *neutral* counterparts *5* and **6,** ylide ion **1** is both thermochemically and kinetically more stable than **216].** In charge stripping  $(CS)$  experiments<sup>[7]</sup>, **1** can be oxidized in the gas phase to its corresponding dication  $H_2C-OH_2^{2+}$  (3), while CH<sub>3</sub>OH<sup>+</sup> upon gas-phase oxidation does not give rise to a detectable dication CH<sub>3</sub>OH<sup>2+</sup> (4)<sup>[8]</sup>. Similarly, if CH<sub>2</sub>OH<sub>2</sub><sup>+</sup> (1) is subjected to a neutralization-reionization  $(NR)^{[9]}$  experiment, in line with the theoretical  $+$ predictions<sup>[6d]</sup>, no recovery signal for the *neutral* ylide  $H_2C-OH_2$  $(5)$  is observed<sup>[10]</sup> in contrast to CH<sub>3</sub>OH<sup>+</sup> which, of course, can be neutralized to methanol. The failure to neutralize  $H_2C-OH_2^+$  is twofold: (i) The carbon-oxygen binding energy of the neutral ylide **5** is quite small (< 10 kcal/mol) and (ii) the large difference in equilibrium geometries of the charged and neutral species implies that vertically neutralized ylides will be formed with considerable excess

Scheme 1



 $(X = H, D)$  gives rise to  $HCNCX<sub>2</sub><sup>+</sup>$ , which can be successfully neutralized in a beam experiment.

energy, thus facilitating rearrangement and/or fragmentation processes.

Here we report our results on the gas-phase generation and characterization of another important ylide, i.e. hydrogen cyanide *N*methylide HCNCH<sub>2</sub> (8) and its radical cation 7. The neutral species



Figure 1. Collisonal activation mass spectra (helium, 80% T) of (a)  $HCNCH<sub>2</sub><sup>+</sup>$ , (b)  $HCNCD<sub>2</sub><sup>+</sup>$ 

*Chem. Ber.* **1993,** *126,* 2353 -2355 *0* VCH Verlagsgesellschaft mbH, D-69451 Weinheim, 1993 O009-2940/93/1010-2353 \$ 10.00+ .25/0

8 is one of the members of the intriguing  $C_2H_3N$  family<sup>[11]</sup>, and in a recent exhaustive study<sup>[2]</sup> its preparation in an argon matrix (10 K), spectroscopic indentification, and its relationship with other  $C<sub>2</sub>H<sub>3</sub>N$  isomers has been described.

Scheme 2



In the context of our earlier work on the identification of the potentially interstellar  $C_2H_3N$  isomer aminoacetylene<sup>[12]</sup>, we also reported on the collisional activation  $(CA)^{\{13\}}$  mass spectra of several  $C_2H_3N^+$  isomers, including also HCNCH<sub>2</sub><sup>+</sup> (7). This species is easily accessible by electron-impact-induced fragmentation of aliphatic isonitriles (Scheme 2), and a typical CA mass spectrum for this ion, generated from 2,2-dimethylpropyl isocyanide, is given in Figure la. In Figure Ib, the CA mass spectrum of the isotopomer  $HCNCD<sub>2</sub><sup>+</sup>$  is reproduced<sup>[14]</sup>.

The interpretation of Figure 1 is straightforward in that  $-$  in addition to the structure-unspecific losses of  $H_x$  ( $x = 1-3$ ) - we observe three diagnostic features one would indeed expect for an  $HCNCX_2^+$  (X = H, D) connectivity: Formations of  $CX_2^+$ , NCX<sub>2</sub><sup>+</sup>, and of the doubly charged ion  $HCNCX<sub>2</sub><sup>2+</sup>$ . This signature is not only different from that of the other  $C_2H_3N^+$  isomers, described



Figure 2. Neutralization-reionization mass spectra (xenon, 80% T/ oxygen 80% T) of (a)  $HCNCH<sub>2</sub><sup>+</sup>$ , (b)  $HCNCD<sub>2</sub><sup>+</sup>$ 

earlier<sup>[12]</sup>, it is also in keeping with the connectivity of ionized hydrogen cyanide N-methylide 7.

Can 7 also be successfully neutralized? As evidenced by the spectra shown in Figure 2, the structure-indicative fragments are indeed retained, when  $HCNCX<sub>i</sub><sup>+</sup>$  ions are subjected to a neutralizationreionization experiment. In addition, we observe a very intense survivor signal. These two observations clearly demonstrate that - in line with the theoretical expectations<sup>[2]</sup> - the neutral  $HCNCX<sub>2</sub>$  (X = H, D) is a stable molecule also in the gas phase. The relatively high abundances of the  $HCNCX_2$  signals in the NR specrelatively high abundances of the HCNCX<sub>2</sub> signals in the NR spectra are no surprise if one takes into account, that  $-$  in contrast to  $H_2O-OH_2$  – no low-energy dissociation paths are available for hydrogen cyanide N-methylide (8). In particular, Maier's calculations<sup>[2]</sup> indicate that the two carbon-nitrogen bonds of 8 are of nearly equal bond length, thus favouring the ylide form **8a**  in comparison to the N-methylide form **8b,** which would be expected to fall apart more easily to HCN and  $CH<sub>2</sub>$ .

$$
HC = N = CH2 \t HCN - CH2
$$
  
8a 8b

The generous financial support of our work by the *Deutsche Forschungsgemeinschaft* and the *Fonds der Chemischen Industrie* is appreciated. H. S. is grateful to the *Alexander uon Humboldt Foundation* for a Max Planck Research Award which provides the financial background for the collaboration with his Israeli partner, Professor *C. Lifshitz,* from the Hebrew University of Jerusalem. We thank Professor G. *Maier* for providing us with a preprint of his recent work on  $C_2H_3N$  isomers in argon matrices.

- **[21** G. Maier, C. Schmid, H. P. Reisenauer, E. Endlein, D. Becker, J. Eckwert, B. A. Hess, Jr., L. J. Schaad, *Chem. Ber.* 1993, 126,  $2337 - 2352$ ; preceding paper.<br><sup>[3]</sup> For a definition, see: L. Radom, W. J. Bouma, R. H. Nobes, B.
- F. Yates, *Pure Appl. Chem.* 1984, 56, 1831.
- **14] L4a1** H. Schwarz, *Nachr. Chem. Tech. Lab.* 1983,31,451. **14b1** H. Schwarz, *Shitsuryo Burseki* 1984, 32, 3. - *Ikl* M. A. Baldwin, *Schwarz, Shitsuryo Burseki* 1984, 32, 3. - <sup>[4c]</sup> M. A. Baldwin, *Specialist Periodical Report, Mass Spectrom.* 1986, 8, 34. - <sup>[4e]</sup> S. Hammerum, *Mass Spectrom. Rev.* 1988, 7, 123. - <sup>[4e]</sup> G. Bouch-Hammerum, *Mass Spectrom. Rev.* **1988**, 7, 123. - <sup>[4e]</sup> G. Bouch-<br>oux, *Mass Spectrom. Rev.* **1988**, 7, 1, 203. - <sup>[40]</sup> K. M. Stirk, L. K. M. Kiminkinen, M. I. Kenttamaa, *Chem. Rev.* 1992,92,1649.
- 
- <sup>[5]</sup> D. G. Morris, Surv. Prog. Chem. 1983, 10, 189.<br>
<sup>[6] [6a]</sup> W. J. Bouma, R. H. Nobes, L. Radom, *J. Am. Chem. Soc.*<br> **1982**, 104, 2929. <sup>[6b]</sup> W. J. Bouma, J. K. MacLeod, L. Radom, *J. Am. Chem. Soc.* **1982**, 104, 1. *Am.* Chem. 350, 1982, 104, 2550.<br>Lossing, J. K. Terlouw, P. C. Burgers, J. *Am. Chem. Soc.* 1982, 104, 2931. – <sup>[6d]</sup> B. F. Yates, W. J. Bouma, L. Radom, J. *Am. Chem. SOC.* 1987, f09, 2250. '
- <sup>[7]</sup> W. Koch, F. Maquin, D. Stahl, H. Schwarz, *Chimia* 1985, 39, 376.
- [\*I F. Maquin, D. Stahl, A. Sawaryn, P. v. R. Schleyer, W. Koch, G. Frenking, H. Schwarz, J. *Chem. SOC., Chem. Commun.* 1984,
- 504.<br><sup>[9]</sup> Reviews: <sup>[9a]</sup> J. K. Terlouw, P. C. Burgers, B. L. M. van Baar, T. Weiske, H. Schwarz, *Chimia* **1986**, 40, 357. <sup>[9b</sup>] C. Wesde-T. Weiske, H. Schwarz, *Chimia* 1986, 40, 357. - <sup>*9bl*</sup> C. Wesde-<br>miotis, F. W. McLafferty, *Chem. Rev.* 1987, 87, 485. - <sup>*9bl*</sup> J. K. Terlouw, H. Schwarz, *Angew. Chem.* **1987**, 99, 829. *Angew. Chem. Int. Ed. Engl.* **1987**, 26, 805. - <sup>[9d]</sup> H. Schwarz, *Pure Appl. Chem. Int. Ed. Engl.* **1987**, 26, 805. - <sup>[9d]</sup> H. Schwarz, *Pure Appl. Chem.* **1989**, 61, 685. - <sup>[9e]</sup> J. L. Holmes, *Mass Spectrom. Rev.* Chem. 1989, 61, 685. - <sup>[92]</sup> J. L. Holmes, *Mass Spectrom. Rev.* 1989, 8, 513. - <sup>[91</sup> J. K. Terlouw, *Adv. Mass Spectrom.* 1989, 11, 1989, 8, 513. - <sup>[96</sup> J. K. Terlouw, *Adv. Mass Spectrom.* 1989, 11, 984. - <sup>*P81*</sup> F. W. McLafferty, *Science* 1990, 247, 925. - <sup>*194*</sup> F. W. McLafferty, Int. J. Mass Spectrom. Ion Processes 1992, *118/f19,* 221.
- ['I C. E. C. A. Hop, J. Bordas-Nagy, J. L. Holmes, J. K. Terlouw, *Mass Spectrom.* **1988**, *23*, **155.** *pp. 1981*, 93, 425; *Angew. html* **1981**, 93, 425; *Angew. Pp. 1981*
- Org. Mass Spectrom. **1988**, 23, 155.<br><sup>[11a]</sup> H. Bock, B. Solouki, Angew. Chem. **198**1, 93, 425; Angew.<br>Chem. Int. Ed. Engl. **1981**, 20, 427. <sup>[11b</sup>] T.-K. Ha, M. T. Chem. Int. Ed. Engl. 1981, 20, 427. - <sup>[116]</sup> T.-K. Ha, M. T.<br>Nguyen, *J. Mol. Struct. (THEOREM)* **1982**, *87*, 355. - **[116]** H. Bock, R. Dammel, S. Aygen, J. *Am. Chem. SOC.* 1983, *105,* 7681.

<sup>[</sup>'I Dedicated to Professor *Hans Bock,* Frankfurt, on the occasion of his 65th birthday.

- **'Itd1** L. L. Lohr, Jr., M. Hanamana, K. Morokuma, *J. Am. Chem. Soc.* **1983, 105,** 5541. - **['Ie1 S.** Saeb, L. Farnell, N. V. Chem. Soc. 1983, 105, 5541. -- <sup>[11e]</sup> S. Saeb, L. Farnell, N. V. Riggs, L. Radom, *J. Am. Chem. Soc.* 1984, 106, 5047. -- <sup>[110</sup>] M. Bogey, J.-L. Destombes, **J.-M.** Denis, J.-C. Guillemin, *J. Mol. Spectrosc.* **1986,** *115,* **1.** - H. Bock, R. Dammel, *Angew.*  Chem. 1987, 99, 518; Angew. Chem. Int. Ed. Engl. 1987, 26, 504.<br>-- <sup>[11h</sup>] H. Bock, R. Dammel, *Chem. Ber.* 1987, 120, 1971. --<br><sup>[11i]</sup> J.-C. Guillemin, J.-M. Denis, M.-C. Lasne, J.-L. Ripoll, *Tetrahedron* **1988,** *44,* 4447.

- [I2] B. v. Baar, W. Koch, C. Lebrilla, **J.** K. Terlouw, T. Weiske, H. Schwarz, *Anyew. Chem.* **1986,** *98,* **834;** *Angew. Chem. Znt. Ed. Engl.* **1986,** *25,* 827.
- **[I3]** Review: **K.** Levsen, H. Schwarz, *Angew. Chem.* **1976,** *88,* 589; *Angew. Chem. Int. Ed. Engl.* **1976,** *15,* 509.
- <sup>[14]</sup> The mass-spectrometric experiments were performed using our modified ZAB-HF machine which is of *BEBE* configuration (*B* stands for magnetic and *E* for electric sector) (ref.<sup>[15]</sup>).  $C_2H_3N^+$ and  $C_2HD_2N^+$  ions were generated by 70-eV ionization of  $(CH_3)_3\overline{CCH_2}NC$  and  $(CH_3)_3\overline{CCD}_2NC$ , respectively, employing the following ion-source conditions: ion-source temperature 200°C; trap current 100 μA; repeller voltage 30 V; acceleration voltage 8 kV; mass resolution  $m/\Delta m = 3000$ . In the NRMS

experiment, a beam of  $B(1)E(1)$  mass-selected  $C_2H_3N^+$ , having 8 keV translational energy, was neutralized in the first cell of a differentially pumped tandem collision cell by colliding it with xenon (80% transmission, T). Unreacted ions were deflected away from the beam of neutral species by putting a voltage on a deflector electrode which is situated between the two collision cells. Subsequent reionization of the beam of neutral  $C_2H_3N$  is brought about in a second collision cell by collision with oxygen (80% T). The mass spectra of the resulting ions were recorded by scanning B(2). The *minimal* lifetime *t* (identical with the transit time from collision cell I to cell **11)** in the present experiments is  $> 10 \mu s$ . In the CA experiments, helium was used as a collision partner **(80% T).** In order to improve the **S/N** ratio, signal averaging techniques were used by accumulating *25* - 50 scans and on-line processing the data with the AMD-Intectra data system.

**[''I [lSa1** R. Srinivas, D. Siilzle, W. Koch, C. H. DePuy, **H.** Schwarz, *J. Am. Chem. Soc.* **1991,** *113,* 5970. - **[lSb1** R. Srinivas, D. K. **Bohme, D. Sülzle, H. Schwarz,** *J. Phys. Chem.* **<b>1991**, *95*, 9836. - <sup>[15c]</sup>, **R. Siilzle, T. Weiske, H. Schwarz,** *Int. J. Mass* $\sum_{n=1}^{\infty}$  **R. Srinivas, D. Sülzle, T. Weiske, H. Schwarz,** *Int. J. Mass Spectrom. Zon Processes* **1991,** *107,* 369.

[239/93]